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RECENT PROGRESS IN OLIGORIBONUCLEOTIDE SYNTHESIS

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Abstract. The usefulness of the p-nitrophenylethylsulfonyl (NPES) group for 2'-OH protection in oligoribonucleotide synthesis is further investigated. The difficulties of uridine protection are discussed and the p-cyanophenylethyl (CPE) group introduced as a versatile new O⁴-blocking group.

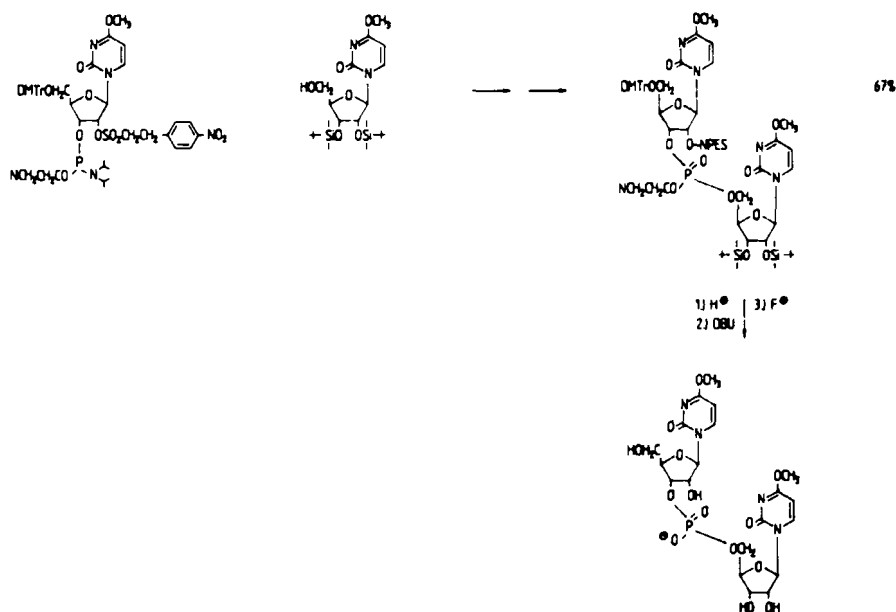
The most significant progress in nucleic acid chemistry in recent years is undoubtedly the development of the effective and convenient phosphoramidite approach^{1,2} for the rapid synthesis of oligo- and poly-deoxyribonucleotides of defined sequence. Less successful, however, was so far the application of the analogous methodology in the oligoribonucleotide series due to the obvious problems concerned with the protection of the additional 2'-OH group. The correct choice of this protecting group, which should remain intact until the final deblocking steps, can be regarded as the crucial point of the whole strategy of a successful oligoribonucleotide synthesis in solution and especially in an automated process on a solid support. A number of protecting groups such as the tetrahydropyranyl-, tetrahydrofuranyl-, 4-methoxytetrahydropyranyl-, 1-(2-chloro-4-methylphenyl)-4-methoxypiperidin-4-yl, 3-methoxy-1,5-dimethoxycarbonyl-pent-3-yl-, benzyl, 2-nitrobenzyl, 4-methoxybenzyl, 3,4-dimethoxybenzyl and tert.butyl dimethylsilyl group have been applied and investigated, but so far no satisfactory solution of the com-

plex problem could be achieved despite the latter blocking group shows some promise³.

Recently we introduced the p-nitrophenylethylsulfonyl (NPES)⁴⁻⁶ group for 2'-OH protection due to the fact that sulfonates at the C-2 position of sugars are generally less reactive in nucleophilic displacement reactions⁷, do not show acyl migration but can be removed in the case of the NPES group by a β -elimination process. Detailed studies revealed that the use of the NPES groups in combination with the p-nitrophenylethoxycarbonyl (NPEOC) group for aglycon protection works very well, whereas phosphate protection has to be performed by the more labile β -cyanoethyl group instead of the p-nitrophenylethyl (NPE) group⁸. The chemical stabilities of the various protected functions indicate that the deblocking sequence has to follow strictly the order 1) phosphate, 2) 2'-hydroxy group and 3) aglycon deprotection, otherwise side reactions occur, which involve the cleavage of the internucleotidic linkage or give rise to anhydronucleoside formation.

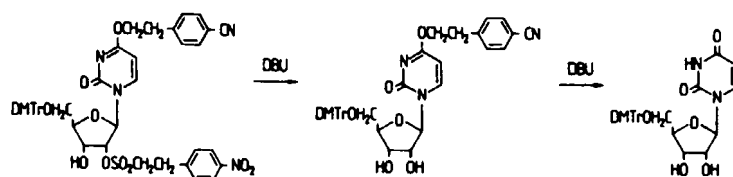
The fully blocked dimers 5'-O-dimethoxytrityl-N⁶-p-nitrophenylethoxycarbonyl-2'-O-p-nitrophenylethylsulfonyl-adenylyl-[3'-(β -cyanoethyl)-5']-N⁶,N⁶,2',3'-tetrabenzoyl-adenosine⁴ and 5'-O-dimethoxytrityl-N⁴-p-nitrophenylethoxycarbonyl-2'-O-p-nitrophenylethylsulfonyl-cytidylyl-[3'-(β -cyanoethyl)-5']-2',3'-di-O-tert.butyl-dimethylsilyl-N⁴-p-nitrophenylethoxycarbonyl-cytidine⁵ respectively showed the expected clean deprotection on subsequent treatment by 1) H⁺, 2) DBU, 3) TBAF and 4) DEAE-Sephadex work-up to give ApA and CpC in an isolated yield of 80 % and 79 % respectively.

More difficulties, however, have been encountered in the uridine series. The use of an unprotected aglycon moiety gives rise to anhydro nucleoside formation since DBU generates an anion which facilitates intramolecular attack of the O² at the C-2' center. O⁴-Protection by the p-nitrophenylethyl group can also not be recommended since the deprotection rates of the NPES and NPE group are not so different that a unique reaction sequence is guaranteed.



Model studies with the fully protected 5'-O-dimethoxytrityl-0⁴-methyl-2'-O-p-nitrophenylethylsulfonyl-uridylyl-[3'-(β-cyanoethyl)-5']-2',3'-di-O-tert-butylidimethylsilyl-0⁴-methyl-uridine, which was synthesized from the corresponding monomeric building blocks by the phosphoramidite approach, revealed a clean deblocking pattern and indicated this way the importance of a relatively stable 0⁴-protecting group if the NPSE group is in use.

In order to develop a new 0⁴-protecting group for uridine, which is more stable than the NPE group but will also be cleaved by a β-elimination process, we draw our attention to the p-cyanophenylethyl (CPE) group⁹, in which the cyano function activates the β-hydrogen atoms much less than the nitro group. After synthesis of the 0⁴-p-cyanophenylethyl-5'-O-dimethoxytrityl-2'-O-p-nitrophenylethylsulfonyl-uridine we noticed that DBU treatment at room temp. removes only the NPES group, whereas the p-cyanophenylethoxy function stays intact. Cleavage by DBU takes place at elevated temp. of about 50°C and shows under these conditions no side reactions.



Thereafter the fully protected dimer 0⁴-p-cyanophenylethyl-5'-O-dimethoxytrityl-2'-O-p-nitrophenylethylsulfonylethyl-uridylyl-[3'-(β-cyanoethyl)-5']-0⁴-p-cyanophenylethyl-2',3'-di-O-tert.butyl-dimethylsilyl-uridine has then been synthesized for further deblocking studies. Treatment first with acid and then with DBU at room temp. led to the partly protected phosphodiester intermediate, which was converted into UpU in high 80 % yield by DBU at 50°C and fluoride ion respectively.

Furthermore we also studied the stability of various substituted phenylethylsulfonylethyl groups towards the influence of DBU in the adenosine series. Introduction of the 4-cyanophenylethylsulfonylethyl, 2-chlorophenylethylsulfonylethyl and 4-methoxyphenylethylsulfonylethyl group into the N⁶-p-nitrophenylethoxycarbonyl-3',5'-O-tetra-isopropyl-disiloxan-1,3-diyl-adenosine led to the fully protected analogues. DBU treatment at room temp. and 50°C respectively revealed that only the NPEOC group was eliminated, whereas the sulfonylethyl residues remained stable under these conditions. It was noticed, however, that treatment of the 2'-O-4-cyanophenylethylsulfonylethyl (CPES) derivative with fluoride ion does not only cleave the cyclic silyl blocking group but also the PCES group at a much faster rate than the relatively stable N⁶-NPEOC protecting group. The 2-chloro- and the 4-methoxyphenylethylsulfonylethyl group respectively again did not show any tendency to cleavage under the influence of fluoride ions. The different reactivities of the β-eliminating blocking groups towards various bases will be subject of future investigations.

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